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It is established that in the presence of small quantities of antimony (0.05 mg/l) and cobalt (20 mg/l) during the electrolytic deposition of zinc an increase in polarisation is observed. When cobalt is present in solutions containing 0.2 mg/l of antimony, the process is determined by the presence of antimony. Antimony, introduced into the electrolyte either as potassium antimonyl tartrate (tartar emetic) or as antimony sulphate, has the same influence on the course of the polarisation curves. It is suggested that small (Chemistry--Physical, TT, v. 5, no. 7) (over)

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